

11th October 2014

Marks: 20

Time: 1 hour

~~Q1.~~ The coordination complexes, **A** and **B** with M(II) ion are represented by $t_{2g}^4 e_g^2$ and $e^3 t^3$ electronic configurations respectively. Identify **A** and **B** and calculate crystal field stabilization energy of each complex (2 marks)

~~Q2.~~ Photochemical reaction of a metal carbonyl, $M(CO)_5$ (**A**) results in the formation of a bimetallic complex, $M_2(CO)_9$ (**B**) with the elimination of CO. Assuming that both **A** and **B** follow EAN rule, Identify **M** and draw the structure of the complex, **B** (2 marks)

~~Q3.~~ State whether the following statements are **TRUE** or **FALSE**? Justify your answer in ONE or TWO lines.

- (i) Octahedral complexes of Co(II) metal ion in both low and High spin states show significant tetragonal distortion.
- (ii) The ionic radii of $[Fe(CN)_6]^{4-}$ is smaller than that of $[Fe(H_2O)_6]^{2+}$
- (iii) The function of both Hemoglobin and myoglobin is identical in biological systems.
- (iv) In the infrared spectra, the C-O stretching band of $Cr(CO)_6$, $[V(CO)_6]^-$ and $[Ti(CO)_6]^{2-}$ appears at 1750, 1860 and 2000 cm^{-1} , respectively. (4x2 = 8 marks)

~~Q5.~~ The oxidative addition of Et_3SiH on a square planar complex, $Ir(CO)(Br)(Ph_2PCH_2CH_2PPh_2)$, (**A**) gives an Ir(III) complex (**B**). The complex **B** undergoes reductive elimination subsequently with the elimination of HBr and forms a square planar complex (**C**). Calculate the valence electrons of (**B**) and (**C**) and draw their structures. (Hint: Ir belongs to Co group) (2 marks)

Q6. State True/False with a one sentence reason. ($1/2 \times 6 = 3$ marks)

- ~~a.~~ For the $v = 17$ harmonic oscillator wavefunction, there is a node at the origin.
- ~~b.~~ For the harmonic oscillator the potential energy can be greater than the total energy.
- c. The spacing between rotational levels remains constant as the quantum number l increases.
- d. The \hat{L}^2 eigenvalues are degenerate except for $l = 0$.
- ~~e.~~ If two operators \hat{A} and \hat{B} commute they have a simultaneous set of eigenfunctions
- ~~f.~~ The value zero is never allowed as an eigenvalue.

Q7. Plot the radial distribution function ($r^2 R^2$) as a function of r for the wavefunction given by $\frac{1}{4(2\pi)^{1/2}} \left(\frac{1}{a_0}\right)^{5/2} r e^{-r/2a_0} \sin \theta \cos \phi$. Determine the most probable radius for this wavefunction. Determine the three quantum numbers associated with this wavefunction. What is the degeneracy associated? (1+1+1/2 +1/2 = 3 marks)